

In situ determination of NO and N₂O from cow-urine applied to a pasture soil under summer conditions

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Abstract

A field experiment was performed during summer using bovine urine at 0, 500 and 1000 kg N/ha to study the effect of urine-N rate on NO_x (NO + NO₂) and N₂O emissions. A Templeton silt loam soil (pH 5.2) was used (Typic Immature Pallic, New Zealand soil classification). The experimental design was a randomized complete block. The main objective of the study was to obtain NO_x data from pasture under bovine urine patches for New Zealand conditions. The experiment was run for 3 months with measurements of some pH, moisture, temperature and inorganic-N and gas (NO_x and N₂O) fluxes, taken over this time. It was found that the emission factors after 3 months for NO-N in the summer were 0.15 and 0.20% of the urine-N applied for the 500 and 1000 kg N/ha rates respectively while the respective N₂O-N fluxes were 0.14 and 0.16%. Multiple regression analyses were used to predict the best regression model for NO-N flux at 500 and 1000 kg N/ha treatment. The best regression model was obtained at 1000 kg N/ha using the soil moisture and the rate of NH₄⁺-N depletion as significant variables. In contrast, the best regression model for N₂O-N fluxes was obtained at 500 kg N/ha treatment using surface soil pH, logNO-N, N₂O-N:NO-N ratio and soil moisture as significant variables.

Key Words

Nitric oxide (NO), nitrous oxide (N₂O), bovine urine, season, multiple regressions.

Introduction

Anthropogenic endeavours have led to changes in the global nitrogen cycle. As a consequence gaseous emission of nitric oxide (NO), nitrogen dioxide (NO₂) (hereafter referred to as NO_x), and nitrous oxide (N₂O) have increased. Scientific interest in these gas emissions from agricultural soils has increased due to their adverse environmental effects, both globally and locally. Nitrous oxide is a gas that contributes to greenhouse warming and also a gas that can undergo oxidation in the stratosphere to form NO_x which in turn can participate in photochemical processes. These processes have adverse effects on stratospheric ozone (O₃) concentrations. The emission of NO_x from soils is of significance due to its potential effect on tropospheric O₃ formation. Increasing levels of NO_x can adversely affect air quality, which has implications for the health of humans, plants and animals. The economic loss of N fertilizer as NO_x gas may represent 1-10% of the N applied (Veldkamp 1997). These NO_x gases are also responsible for the atmospheric acidity and nitrogen to downwind ecosystems. In soils, the microbial processes responsible for NO_x and N₂O gas emissions are known to be nitrification and denitrification, while abiotic production also occurs to a lesser and unknown extent (Venterea and Rolston 2000). The majority of studies on NO_x gas emissions have focused on fertilizer applications with very few studies (ca. 5) examining the potential for NO_x emissions from urine patches in grazed pastures. Studies that have examined NO_x gas emissions from urine have used unrealistically low rates of N, for New Zealand systems, or concentrated on soils with high moisture contents where NO_x gas emissions might be expected to be low. Of these, none have examined urine-N rates and/or seasonal effects on urine-N NO_x fluxes. There are no in situ studies in New Zealand temperate pastures that have examined the potential for NO_x emissions from bovine urine.

The main objectives of this experiment were: To obtain the NO_x emission data from pastures under two rates of bovine urine, under New Zealand summer conditions; To examine the relationships between NO and the N₂O fluxes and between NO fluxes and the measured soil variables from urine treated soils.

Methods

A field experiment was conducted in which urine, collected from the Lincoln University dairy farm, with urinary-N concentration of 10.5 g N/L was applied to experimental plots. Experimental plots were either gas chamber plots (0.029 m²) or soil sampling plots (0.16 m²). The experimental design consisted of three

treatments (500 kg N/ha, 1000 kg N/ha and a control) replicated 3 times. Gas sampling for NO_x and N_2O was conducted periodically for nearly 3 months using chemiluminescence and gas chromatography methods respectively. The soil NH_4^+ -N and NO_3^- -N concentrations were measured concurrently with the gas sampling events over a depth of 0–7.5 cm but the soil NO_2^- -N concentration was determined for 3 soil depths of 0–2.5, 2.5–5.0 and 5.0–7.5 cm.

Results

Results showed that urine-N rate significantly affected the inorganic-N concentrations. Maximum NH_4^+ -N and NO_3^- -N concentrations were recorded at the highest urine-N rate (1000 kg N/ha) followed by the subsequent urine-N rate (500 kg N/ha) (Figures 1 and 2). The soil NO_2^- -N concentrations did not show any significant differences between depths due to treatments. When data from the 3 soil depths were pooled the soil NO_2^- -N concentrations were negatively correlated with soil pH ($r = -0.22$, $P < 0.01$). The NH_4^+ depletion rate was 1.8 fold higher in the 1000 kg N/ha than the 500 kg N/ha treatment. No significant differences in terms of surface soil pH were observed in urine-N treated soils. However, the urine-N treated soils had higher soil pH values than the control treatment. Fluxes of NO-N varied over time and with urine-N rate (Figure 3). Maximum NO-N fluxes were observed at the high urine-N rate of 1000 kg N/ha treatment. Cumulative NO-N losses for the 500 and 1000 kg N/ha treatments were 0.15 and 0.20% of the applied urine-N respectively. NO-N fluxes were positively correlated with surface soil pH, NO_3^- -N, NH_4^+ -N, net ammonium depletion rates and net nitrate accumulation rates. Urine-N rate also significantly affected the N_2O -N fluxes over time (Figure 4). Maximum cumulative N_2O -N fluxes (0.16%) were recorded at 1000 kg N/ha followed by the 500 kg N/ha treatment (0.14%). When the N_2O -N flux data from all sampling dates were pooled N_2O -N fluxes were significantly correlated with surface soil pH and NH_4^+ -N concentration. Soil temperature and WFPS during the experimental period ranged from 14–30°C and 17.5–35.6% respectively.

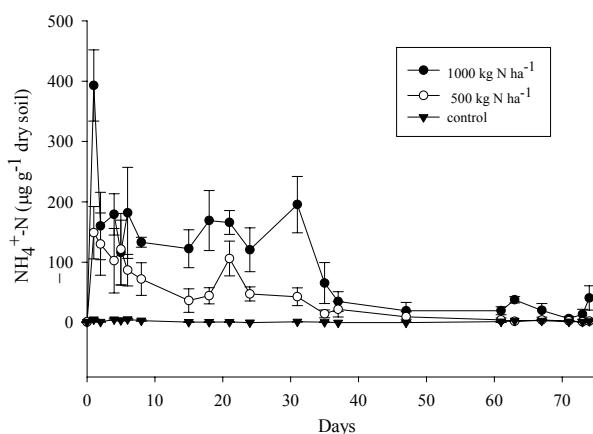


Figure 1 Soil NH_4^+ -N concentrations over time ($n = 3$, error bars are \pm s.e.m.).

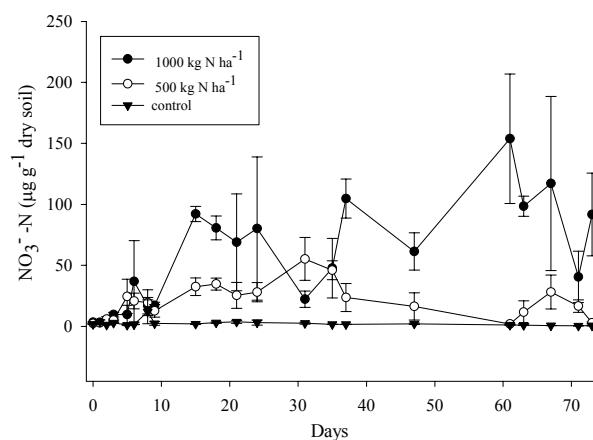


Figure 2. Soil NO_3^- -N concentrations over time ($n = 3$, error bars are \pm s.e.m.).

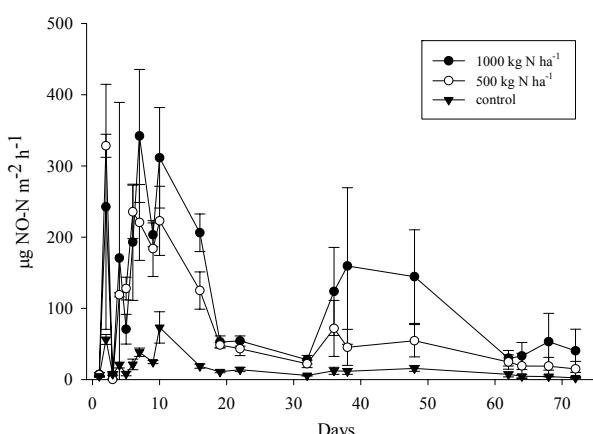


Figure 3. NO-N fluxes over time following urine treatment application ($n = 3$, error bars \pm s.e.m.).

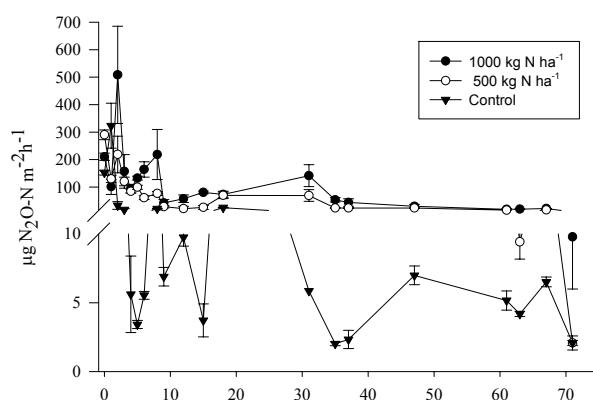


Figure 4. Soil N_2O -N fluxes over time following urine treatment application ($n = 3$, error bars \pm s.e.m.).

Conclusion

The cumulative NO-N flux from the 500 kg N/ha treatment was very comparable with the maximum flux attained in the other field studies where bovine urine has been applied under summer conditions. No other in situ studies have been performed using 1000 kg urine-N/ha and the current study represents the first at this rate. No previous in situ urine studies have attempted to examine the relationships between the net NO-N flux, the soil NO_2^- -N pool and soil pH. Despite the presence of NO_2^- -N in the soil, no relationship was found between theoretical HNO_2 concentrations and the NO-N flux. Thus net NO-N emissions are believed to have been predominately from microbial sources. Further evidence of biological production of NO was seen with the strong relationship between $\log\text{N}_2\text{O}$ -N and $\log\text{NO}$ -N emissions. The gaseous N losses observed in the present study were not significant in agronomic terms, but assist in assessing the contribution of such losses to the atmospheric budget for NO_x . Further studies are required to elucidate in-situ production mechanisms of NO and the relationship to N_2O in the urine patch.

References

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